

## Matter And Intermolecular Forces Concept Review Answers

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### Matter And Intermolecular Forces Concept

As was the case for gaseous substances, the kinetic molecular theory may be used to explain the behavior of solids and liquids. In the following description, the term particle will be used to refer to an atom, molecule, or ion. Note that we will use the popular phrase "intermolecular attraction" to refer to attractive forces between the particles of a substance, regardless of whether these ...

### 11.1 Intermolecular Forces - Chemistry Fundamentals

Watch different types of molecules form a solid, liquid, or gas. Add or remove heat and watch the phase change. Change the temperature or volume of a container and see a pressure-temperature diagram respond in real time. Relate the interaction potential to the forces between molecules.

### States of Matter - Atomic Bonding | Interaction Potential ...

These forces are weaker than intermolecular forces. These forces are responsible for the liquids, solids and solutions state of any compound. Some common types of intermolecular forces are London dispersion, dipole-dipole, Hydrogen bonding and ion-ion force. The order of strength of these intermolecular forces is given below.

### London Dispersion Forces - Definition, Examples, Formula ...

A force is a push or pull acting upon an object as a result of its interaction with another object. There are a variety of types of forces. Previously in this lesson, a variety of force types were placed into two broad category headings on the basis of whether the force resulted from the contact or non-contact of the two interacting objects.

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