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Chemistry Notes for class 12 Chapter 2 Solutions Solution is a homogeneous mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution. Solute and Solvent

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Homogeneous mixtures of two or more than two components are called solutions. In the year 1986, Francois Marté a French chemist proposed a law.

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Class 12 Chemistry Revision Notes for Chapter 2 - Solutions

There are three types of solutions:- True solutions:- the solution in which the size of solute particles is less than 10^{-9} m or 1 nm ($< 10^{-9}$ m). Example. solution of sugar in water. Colloidal solution:- the solution in which the size of solute particles is in between 10^{-9} to 10^{-6} m are called colloids. Example. Water in milk.

CBSE Class 12 Chemistry Notes: Surface Chemistry ...

Revision Notes on Solution: A solution is a homogeneous mixture of two (or more) substances, the composition of which may vary between certain limits. A solution consisting of two components is called binary solution. The component which is present in large quantity is called solvent and the component which is small in quantity is called solute.

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updating....). These notes are aligned with our Channel's Video Lectures. At times, you may find some difference between video lecture & Pdf Handwritten Notes. Kindly note that some of these notes are not written by us and they are borrowed from student community.

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CBSE Class 12th Chemistry - Chapter 5 Surface Chemistry
Important Notes are available. Topics covered are: Colloids,
Classification of colloids, Physical state of dispersed phase and
dispersion ...

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