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substance interferes with hydrogen bonding in water. surfactant. dissolving medium known as a solution. solvent. a homogeneous solution of water and another substance is called a(n)

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To show how you can predict whether substances are soluble in water.

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organic) compounds. a. sodium chloride, NaCl, in table salt inorganic b. hexane, C<sub>6</sub>H<sub>14</sub>, in gasoline organic c. ethyl butanoate, CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>, in a pineapple organic d.

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reference see “Chemistry: An Atoms-focused Approach” by Gilbert, Kirss, and Foster. \*In addition to reviewing this study guide, you should i) consult the Chapter Objectives posted on Blackboard and ii) PRACTICE PROBLEMS (Old exams, mock exams, end of chapter problems, etc.)

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nucleus.

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chemistry chapter 15 vocabulary.

surface tension. surfacant. aqueous solution. solvent. an inward force that tends to minimize the surface area of a l.... any substance that interferes with the hydrogen bonding betwee.... water

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that contains dissolved substances. the dissolving medium in a solution.

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