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Chemistry 1451 Chapter 4 Homework / Study Guide a.  $1.50 \times 10^{25}$  molecules  
24.9 mol b.  $2.58 \times 10^{21}$  molecules  
0.00428 mol c.  $7.9 \times 10^{24}$  molecules  
13.1 mol 6. Using electronegativity values from the table below, classify the bond formed between each pair of elements as polar covalent or ionic.

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